Copyright is owned by the Author of the thesis. Permission is given for a copy to be downloaded by an individual for the purpose of research and private study only. The thesis may not be reproduced elsewhere without the permission of the Author.

# CADMIUM MANAGEMENT IN NEW ZEALAND'S HORTICULTURAL SOILS

A thesis presented in partial fulfilment of the requirements for the degree of Master of Environmental Management

Massey University, Palmerston North, New Zealand



Hadee Thompson-Morrison 2017

Copyright is owned by the Author of the thesis. Permission is given for a copy to be downloaded by an individual for the purpose of research and private study only. The thesis may not be reproduced elsewhere without the permission of the Author.

#### Abstract

Cadmium (Cd) is a heavy metal trace element which presents risks for the horticultural industry in New Zealand (NZ). This element is added to soils through phosphate fertiliser application, and once there may be available for plant uptake and food chain transfer. When food products exceed international standards for Cd concentrations, these products may be excluded from international markets upon which NZ relies to maintain its economy. This presents a reputational risk for NZ's horticultural exports. Soil pH and organic matter (OM) content are the two key drivers influencing Cd's bioavailability, and field trials are currently being undertaken in four horticultural sites throughout NZ – Pukekawa, Manawatu, and two adjacent sites at Lincoln – to test the efficacy of the use of lime and compost amendments to influence these soil variables and thus reduce Cd plant uptake from soils. Potatoes are grown at all sites while Lincoln also includes wheat. This research aimed to characterise these soils, including total and plant-exchangeable Cd concentrations, pH, OM content, cation exchange capacity, total and plant-exchangeable Zn concentrations, aluminium and iron oxide content, total phosphorus and total nitrogen content. Findings indicated that total Cd concentrations varied among sites, with the highest (0.52 mg kg<sup>-1</sup> 1) at Pukekawa, followed by Manawatu (0.26 mg kg<sup>-1</sup>) and Lincoln Wheat and Potato sites (both 0.13 mg kg<sup>-1</sup>). Exchangeable Cd concentrations were low at all sites (0.01-0.02 mg kg<sup>-1</sup>) indicating little risk of plant uptake from these soils.

The mitigation strategy tested in this work focuses on pH as a key soil variable that can be readily changed to restrict Cd uptake. However, the effectiveness of amendment rates to effect target pH values is dependent on soil chemistry and rates will vary across sites. Incubation experiments were conducted to determine amendment rates for lime and sulphur, and to compare the pH of amended soil in a laboratory situation with the in-field situation. Incubation and field situations were found to be similar, with no significant differences between pH values after a period of 274 days in the incubator and 169 days in the field. The accuracy of the calculated amendment rates at achieving target pH values was assessed with extended incubation experiments. The results here varied between soils, with the sulphur application rate proving more accurate in the Pukekawa soil, however too high for the Manawatu and Lincoln potato soils. The calculated liming application rate similarly resulted in a higher-than-target pH, however after a period of 231-274 days the pH reduced and approached the target value.

A cost-benefit analysis was undertaken to determine the economic viability of the proposed mitigation strategy at each potato site. Results proved the strategy to be a viable option, which would remain viable in the face of varying uncertainty and reductions in potato yields. Practical considerations including timing and weather conditions, and compost availability were considered. Implementation of this strategy within NZ's current framework of the Tiered

Fertiliser Management System, which focuses on total rather than exchangeable Cd concentrations, may present difficulties, and thus there is a clear need for risk-based, soil and crop specific guidelines for Cd management within a NZ context.

Considering the apparent difficulties in designing pH amendments strategies, a model to convert pH buffer curve-generated lime application rates which can be derived in as little as 24 hours, to field applicable application rates which target a specific soil pH was developed for the Pukekawa soil. A similar model was not achieved for the Lincoln Wheat soil, and thus the development of such a model is not possible for all soil types. Where possible, the development of this model would be an innovative and useful tool for farmers with which to accurately and quickly determine required lime application rates to achieve a targeted soil pH. This would be of great benefit in the implementation of a Cd mitigation strategy using lime amendments, and would allow greater control over, and management of, soil pH in a horticultural context.

#### Acknowledgements

There are so many people to whom I am grateful for their guidance and support throughout what has been such an enjoyable journey. Firstly, to my supervisor, Associate Professor Chris Anderson, I am grateful for the guidance and insights he afforded me throughout the completion of this degree, and for his consistent encouragement to 'tell the story' that is this thesis and all it encompasses in a way which kept me on track. I am extremely appreciative of the help I have received from my co-supervisor, Dr. Paramsothy Jeyakumar (Jeya), and his patience in teaching me laboratory methods while never failing to provide an encouraging voice.

I extend my sincere thanks to Thangavelautham Geretharan (Gere), who helped me complete the pre-amendment soil analysis and pH incubation work with great patience, and without whose help I would not have completed this work. Thank you for your data collection, analysis, and reporting of results that I have included within this thesis.

To Ian Furkert, who helped me complete my laboratory work including the CEC analysis, and to Glenys Wallace who helped with the phosphorus analysis, I am thankful for the assistance and teachings. I would also like to thank Bob Toes, who helped to elucidate some (what first appeared as) rather perplexing results.

I am profoundly grateful to Emeritus Professor Anton Meister for his help with the costbenefit analysis section of this thesis, for his valuable advice and experience.

To the Cadmium Group, including Jo Cavanagh, Brett Robinson, Niklas Lehto and Michael Yi, I have enjoyed their encouragement and support and have valued this chance to work as part of wider group on such an involved project.

I would like thank the Fertiliser Association of New Zealand for the financial support of this project, without which none of this would be possible. My sincere appreciation also goes to my scholarship providers, Lovell & Berys Clark, the C. Alma Baker Trust and Seed Tech Services for their generous support.

To David Perl, who was of great assistance as both a critical eye, proof reader and friend, thank you for the help you offered me, and for your unfailing encouragement when I needed it. Also to Heather Jameson, where would I be without you? Thank you for your support and care throughout all areas of my life, and for your much appreciated last-minute proof reading.

Most importantly, to my mother, Jo-anne Thompson, who is not here today to see me complete this, I'm forever grateful for the wonderful upbringing she gave me and for never failing to encourage me to chase my dreams and achieve to my highest ability.

Throughout this process my friends and family have been an invaluable source of support and encouragement, and it is to them that I offer my final acknowledgement, thank you all for making this such a wonderful journey.

### Table of Contents

Abstract	II
Acknowledgements	
List of Figures	
List of Tables	xi
List of Appendix Tables	xiv
1 Introduction	
2 Background	
2.1 Cadmium – A Non-Essential Trace Element	2
2.2 Cadmium's Harmful Effects on Plants and M	licroorganisms3
2.3 Soil Characteristics Influencing Cadmium Bi	oavailability4
2.3.1 Total and Exchangeable Cadmium Conto	ent2
2.3.2 Soil pH	5
2.3.3 Organic Matter Content	
2.3.4 Cation Exchange Capacity	S
2.3.5 Zinc	<u>9</u>
2.3.6 Aluminium & Iron Oxides	
2.3.7 Total Phosphorus	
2.3.8 Total Nitrogen	
2.4 Plant Factors Influencing Cadmium Bioavail	ability12
2.4.1 Plant Species	
2.4.2 Plant Cultivar	
2.4.3 Plant Tissue	
2.4.4 Leaf Age	
2.5 Cadmium Accumulation in a New Zealand C	Context
2.5.1 Background	
2.5.2 Risks and Implications of Cadmium Acc	cumulation for New Zealand 18
2.5.3 New Zealand's Cadmium Management S	Strategy 19
2.6 Mitigations for Cadmium Plant Uptake	21
2.6.1 Organic Amendments	21
2.6.2 Lime	25
2.7 Environmental Management of Cadmium in	New Zealand Soils28
2.8 Current Cadmium Research in New Zealand.	29
2.9 Research objectives	30
291 Current Research Aims	3(

3 Site Characterisation – Introduction to Study Farms	32
3.1 Materials and Methods	32
3.1.1 Soils and Sampling	32
3.1.2 Soil Characterisation (Pre-amendment Analysis) Methods:	33
3.2 Results and Discussion	35
3.2.1 Pukekawa	35
3.2.2 Manawatu	38
3.2.3 Lincoln Potato	38
3.2.4 Lincoln Wheat	39
4 Research to Derive Accurate Application Rates to Achieve Target pH Values in Soil	41
4.1 Materials and Methods	41
4.1.1 Soils	41
4.1.2 Lime and Sulphur	41
4.1.3 Soil Buffering Capacity	41
4.1.4 Amendment Rate Determination	42
4.1.5 pH Incubations	43
4.2 Results & Discussion	43
4.2.1 Soil Buffering Capacity	43
4.2.2 Amendment Rates for use in pH Incubations	44
4.2.3 pH Incubation Results	47
4.2.4 Calculation of Field Trial Amendment Rates	50
4.2.5 Extended pH Incubation Results	50
4.2.6 pH: From Incubation to Field	54
5 Cost-Benefit Analysis	58
5.1 Methodology	58
5.1.1 Criteria for Viability	58
5.1.2 The Project Scenario	58
5.1.3 Determination of Costs and Benefits	59
5.1.4 Determination of the Project's Net Present Value	61
5.1.5Sensitivity Analysis	62
5.2 Results	62
5.2.1 Sensitivity Analysis	66
5.3 Environmental Management: How Realistic is this Mitigation Strategy?	69
5.3.1 Soil pH and Potato Scab	69
5.3.2 Compost Supply	69
5.3.3 On-farm Considerations	69

	5.3.4Towards a Key Step: Achieving a Target pH	70
	5.3.5 Recommendations for Implementation	74
6 (	Conclusions	. 75
7F	References	. 77
8 4	Appendices	1
	Appendix A: Personal Communications	1
	Appendix B: Soil Characterisation Data	1
	Appendix C: Soil pH Buffering Capacity and Amendment Calculations	1
	Appendix D: Discount rates and factors for CBA purposes, and equations	1

## List of Figures

Figure 1. Topsoil Cd concentrations throughout NZ as determined by Taylor et al. (2007, p.	
15)	. 17
Figure 2. K <sub>d</sub> values of organic amendments and soils through a range of pH values, produced	by
Al Mamun et al. (2016a)	. 24
Figure 3. Location of horticultural soil sites used. Soil types at each site are detailed in the ma	ιp
legend according to the New Zealand Soil Classification.	.33
Figure 4. Buffer curves of all soils inclusive of pH target range	.48
Figure 5. Pukekawa soil pH after seven days of incubation with aglime, hydrated lime and	
elemental S amendments	.48
Figure 6. Pukekawa soil pH after 10 days of incubation with aglime, hydrated lime and	
elemental S amendments	.48
Figure 7. Manawatu soil pH after seven days of incubation with aglime, hydrated lime and	
elemental S amendments	.48
Figure 8. Manawatu soil pH after 10 days of incubation with aglime, hydrated lime and	
elemental S amendments	.48
Figure 9. Lincoln Potato soil pH after seven days of incubation with aglime, hydrated lime an	d
elemental S amendments	.49
Figure 10. Lincoln Potato soil pH after 10 days of incubation with aglime, hydrated lime and	
elemental S amendments	.49
Figure 11. Lincoln Wheat soil pH after seven days of incubation with aglime, hydrated lime a	
elemental S amendments	.49
Figure 12. Lincoln Wheat soil pH after 10 days of incubation with aglime, hydrated lime and	
elemental S amendments	.49
Figure 13. Manawatu incubation pH at day 58 for aglime*10, elemental S*3 and control	
samples	. 52
Figure 14. Manawatu incubation pH at day 121 for aglime*10, elemental S*3 and control	
samples	. 52
Figure 15. Manawatu incubation pH at day 231 for aglime*10, elemental S*3 and control	
samples	. 52
Figure 16. Lincoln Potato incubation pH at day 101 for aglime*10, elemental S*3 and control	L
samples	. 53
Figure 17. Lincoln Potato incubation pH at day 274 for aglime*10, elemental S*3 and control	Ĺ
samples	. 53
Figure 18. Pukekawa soil incubation pH after 101 days	.56
Figure 19. Pukekawa soil incubation pH after and 274 days	.56

Figure 20. Pukekawa field pH of control, S and lime plots 147 days after amendment
application5
Figure 21. Lincoln Wheat incubation pH of control and aglime*10 samples after 101 days 5
Figure 22. Lincoln Wheat incubation pH of control and aglime*10 samples after 274 days 5
Figure 23. Lincoln Wheat field pH of control and lime amendment plots after 139 days 5
Figure 24. Lincoln Wheat field pH of control and lime amendment plots after 169 days 5
Figure 25. Relationship between buffer curve-generated and field application rates of lime to
achieve a target pH in Pukekawa soil
Figure 26. Relationship between buffer curve-generated and field application rates of lime to
achieve a target pH in Lincoln Wheat soil

### List of Tables

Table 1. Differences in Cd accumulation between types of vegetables as reported by different
authors
Table 2. Summary of mean NZ Cd concentrations for varying land-uses sourced from Cavanagh
(2014)
Table 3. Summary of soil Cd concentrations throughout regions of NZ, as presented in
Cavanagh (2014, p. 8)
Table 4. Soil contaminant standards for Cd determined to protect human health in NZ's NES 19
Table 5. Representation of the Tiered Fertiliser Management System from Cavanagh (2012, p.
1), showing tiers, trigger values, and required management actions
Table 6. Studies assessing the effects of lime on soil and plant Cd, with application rates and
primary observations
Table 7. Mean soil characteristics of the study farms at Pukekawa, Manawatu and Lincoln 37
Table 8. Volumes of 0.2 M NaOH, 0.2 M HCl and deionised H <sub>2</sub> 0 added to beakers of 20 g soil
to create pH buffer curve for each soil site
Table 9. Bulk density and field capacity of each soil
Table 10. Native pH and buffering capacities of incubated soils
Table 11. pH incubation amendments and rates for each site, derived from pH buffer curve data
to achieve a range of target pH values
Table 12. Amendment rates of elemental S or aglime per plot used in field trial work in four
experimental locations
Table 13. Percentage variations from target pH of Lincoln Wheat soil in incubations at day 101
and in field at day 139
Table 14. Factors considered within the CBA and their associated values
Table 15. Cost-benefit analysis of mitigation at the Pukekawa site, detailing benefits from
potato revenue, costs of mitigation, present values of total costs and benefits, the project's net
cash flow and the overall NPV
Table 16. Cost-benefit analysis of mitigation at the Manawatu site, detailing benefits from
potato revenue, costs of mitigation, present values of total costs and benefits, the project's net
cash flow and the overall NPV
Table 17. Cost-benefit analysis of mitigation at the Lincoln site, detailing benefits from potato
revenue, costs of mitigation, present values of total costs and benefits, the project's net cash
flow and the overall NPV65
Table 18. Cost-benefit sensitivity analysis for the Pukekawa site using a potato yield of 50 t ha <sup>-1</sup> ,
detailing benefits from potato revenue, costs of mitigation, present values of total costs and
benefits, the project's net cash flow and the overall NPV67

Table 19. Cost-benefit sensitivity analysis for the Manawatu site using a potato yield of 50 t h	1a¯
, detailing benefits from potato revenue, costs of mitigation, present values of total costs and	Į.
benefits, the project's net cash flow and the overall NPV	. 67
Table 20. Cost-benefit sensitivity analysis for the Lincoln site using a potato yield of 50 t ha <sup>-1</sup>	,
detailing benefits from potato revenue, costs of mitigation, present values of total costs and	
benefits, the project's net cash flow and the overall NPV	. 68
Table 21. pH changes in the Lincoln Wheat soil during field trial work, and their associated	
lime application rates	. 73

## List of Appendix Tables

Table B-1. Pukekawa soil characterisation results	_B-1
Table B-2. Manawatu soil characterisation results	B-1
Table B-3. Lincoln Potato soil characterisation results	B-2
Table B-4. Lincoln Wheat soil characterisation results	B-3
Table C-1. pH buffer curve results for Pukekawa soil	C-1
Table C-2. pH buffer curve results for Manawatu soil	
Table C-3. pH buffer curve results for Lincoln Potato soil	C-1
Table C-4. pH buffer curve results for Lincoln Wheat soil	C-1
Table C-5. Soil buffering capacity and amendment application rate calculation table for Pukekawa and Manawatu soil	
Table C-6. Soil buffering capacity and amendment application rate calculation table for Lincoln Wheat and Lincoln Potato soil	C-3
Table D-1. Discount factors by year for discount rates ranging from 1-10%	D-1